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ANSWER Answer the

following questions in
the space provided. 1.

When an electron is
added to a neutral

atom, energy is. (a)

always absorbed (c)

either absorbed or

released (b) always

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law. lanthanide.

actinide. alkali metals.

The physical and

chemical properties of

the elements are

periodic.... The 14

elements with the

atomic numbers 58-71.

The 14 elements with

the atomic numbers

90-103. The elements

of group 1 of the

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ANSWER Answer the
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When an electron is added to a neutral atom, energy is . (a) always absorbed (c) either absorbed or released (b) always released (d) burned away 2.

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ANSWER Answer the following questions in the space provided. 1.

Consider the neutral atom with 53 protons and 74 neutrons to answer the following questions. 53 a. What is its atomic number? 127 b. What is its mass number? atomic number c. Is the element's position in a

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